

Regiocontrolled Cu^I-Catalyzed Borylation of Propargylic-Functionalized Internal Alkynes

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S Supporting Information

ABSTRACT: Good to excellent reactivity and regiocontrol have been achieved in the Cu^I-catalyzed borylation of dialkyl internal alkynes with bis(pinacolato)diboron. The presence of a propargylic polar group (OH, OR, SAR, SO₂Ar, or NHTs), in combination with PCy₃ as ligand, allowed maximizing the reactivity and site-selectivity (β to the propargylic function). DFT calculations suggest a subtle orbitalic influence from the propargylic group, matched with ligand and substrate size effects, as key factors involved in the high β -selectivity. The vinylboronates allowed the stereoselective synthesis of trisubstituted olefins, while allylic substitution of the SO₂Py group without affecting the boronate group provided access to formal hydroboration products of unbiased dialkylalkynes.

One of the greatest challenges associated with selective functionalization of alkynes via hydrometalation is the control of the regioselectivity.¹ Achieving high regiocontrol generally relies on the use of alkynes with a marked bias in either the size or the electronic characteristics of the acetylenic substituents, such as terminal alkynes, aryl-substituted alkynes, and conjugated enynes.¹ In contrast, simple dialkylalkynes without a strong electronic or steric bias (i.e., alkyl-C \equiv C-alkyl) often lead to unpractical regioisomeric mixtures. These characteristics hold true in hydroboration reactions.^{1,2} The catalytic hydroboration of terminal alkynes with B–H reagents² is regarded as one of the most straightforward methods for accessing vinylboronate reagents, which are highly versatile building blocks endowed with numerous synthetic applications.³ The anti-Markovnikov addition was the usual regiochemical outcome in the direct borylation of terminal alkynes until the very recent work by Hoveyda and co-workers, who devised a NHC-Cu-catalyzed borylation protocol with bis(pinacolato)diboron [B₂(pin)₂], leading to functionalized branched vinylboronates with high α -selectivity.^{4,5}

The regiocontrolled borylation of internal unsymmetrical alkynes represents an even greater challenge because of their lower reactivity and difficult regiocontrol. Indeed, only recently have the first two successful reports on highly regioselective borylation of internal alkynes appeared.^{6–8} Both are elegant examples of Cu^I-catalyzed B₂(pin)₂-borylation of electronically biased internal alkynes, such as 1-aryl-1-alkynes⁶ and 1,3-enynes.⁷ In contrast, a general procedure for the highly regiocontrolled hydroboration of simple dialkylalkynes remains an unmet challenge. Herein we present an operationally trivial,

Cu^I-PCy₃-catalyzed protocol for the regio- and stereocontrolled B₂(pin)₂-borylation of propargylic functionalized dialkylalkynes. The resulting vinylboronates were applied to the stereoselective synthesis of trisubstituted alkenes via cross-coupling reactions. Further elaboration of the borylation products without affecting the boronate group via allylic substitution allowed widening the current structural scope in the synthesis of vinylboronates.

Given the wide chemical versatility of sulfur compounds, we envisioned that propargyl sulfur-containing species might have a neighboring regiocontrolling effect in the Cu^I-catalyzed borylation reaction with B₂(pin)₂. To test this hypothesis, model sulfide **1** was subjected to Cu-catalyzed⁹ borylation under the standard reported conditions¹⁰ (Table 1). In

Table 1. Ligand Screening in the Borylation of Alkyne 1

entry	L	conv (%) ^a	α/β ratio ^a
1	–	15	<2 : >98
2	Xantphos	<2 ^b	–
3	dppf	<2 ^b	–
4	<i>rac</i> -Binap	<2 ^b	–
5	Xphos	18	<2 : >98
6	PPh ₃	50	<2 : >98
7	PCy ₃	>98 ^c (76) ^d	<2 : >98
8	P(<i>t</i> -Bu) ₃	60	25 : 75

^aDetermined by ¹H NMR from the crude mixture. ^bStarting material recovered. ^cReaction time 2 h. ^dYield after chromatography.

accordance with literature precedents,^{4–7} we noticed a strong dependence of the reactivity upon the nature of the ligand. Very poor reactivity (15% conversion), yet a very high β -selectivity, were observed in the absence of ligand (entry 1). Typical bidentate phosphines such as Xantphos,¹¹ dppf, or (\pm)-Binap were totally ineffective (entries 2–4). A scan of the recent literature revealed that the level of reactivity and regiocontrol in addition reactions to alkynes can be optimized by adjusting the steric and electronic properties of monodentate phosphine or NHC ligands.^{4–7,12–14} Xphos¹³ and PPh₃ ligands promoted the β -selective borylation, albeit with unpractical conversions (18% and 50%, respectively, entries 5

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and 6). To our delight, the reactivity was greatly enhanced with the more donating PCy₃ ligand (full conversion after 2 h at rt) while maintaining the excellent β -regiocontrol (entry 7). In contrast, the bulkier P(*t*-Bu)₃ was much less reactive (60% conversion) and regioselective ($\alpha/\beta = 25 : 75$; entry 8).

A variety of propargylic functional groups with various steric and electronic properties provided uniformly good reactivity and very high regiocontrol (generally >98% β -selectivity, Table 2). A potentially coordinating 2-pyridylsulfide group had no

Table 2. β -Borylation in Propargylic-Substituted 2-Butynes

entry ^a	FG	product	α/β ratio ^b	yield (%) ^c
1	S(2-Py)	3	<2 : >98	82
2	SO ₂ Ph	4	<2 : >98	78
3	SO ₂ (2-Py)	5	<2 : >98	80
4	OH	6	<2 : >98	69 ^d
5	OBn	7	<2 : >98	78
6	OPh	8	<2 : >98	79
7	OTBDMS	9	<2 : >98	76
8	OTIPS	10	<2 : >98	82
9	OAc	11	10 : 90	76 ^e
10	NHTs	12	<2 : >98	89
11	CH ₂ OBn	13	12 : 88	72 ^f
12	2-pentyne	14	15 : 85	30 ^f
13	3-heptyne	15	43 : 57 ^g	37 ^f

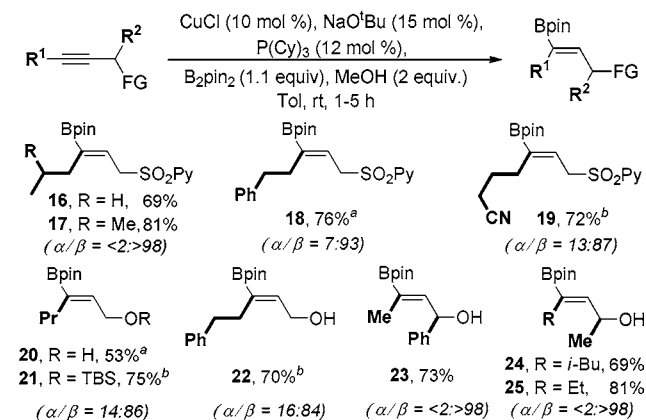
^a0.26 mmol scale in substrate. ^bDetermined by ¹H NMR from the crude mixture. ^cIsolated product after chromatography. ^dReaction carried out in the absence of MeOH. ^eIn pure β -regioisomer after chromatographic separation. ^fYield in the mixture of regioisomers. ^g α/β -site selectivity not assigned.

impact on reactivity or regioselectivity (entry 1), nor did the bulkier and more electrophilic sulfonyl substrates (products 4 and 5, 78–80% yield, entries 2 and 3). Oxygenated propargyl derivatives proved also to be suitable, which is interesting given the availability of propargyl alcohols and the synthetic utility of the resulting allylic alcohols. The parent 2-butyne was successfully transformed into the β -vinylboronate 6, bearing an allylic hydroxyl group (69%, entry 4). A series of propargyl alcohol derivatives delivered the corresponding vinylboronates in reasonable yields (76–82%), irrespective of the substituents attached to the oxygen (Bn, Ph, TBDMS, TIPS, or even the challenging Ac,¹⁴ entries 5–9). Only the acetyl-protected substrate gave a lower regiocontrol ($\alpha/\beta = 10 : 90$). This general trend in both reactivity and regioselectivity was also observed for the *N*-Ts-protected propargylamine (product 11, $\alpha/\beta = <2 : >98$, entry 10). This set of homogeneous results suggests that the observed regioselectivity profile is not derived from direct coordination of the propargylic function to copper. With regard to other types of internal alkynes, the homopropargyl benzyl-protected 3-pentynol derivative provided the corresponding β -vinylboronate with good, albeit significantly lower regiocontrol (13, $\alpha/\beta = 12 : 88$, entry 11) than the propargyl derivative (entry 5). The borylation of a non-functionalized internal alkyne such as 2-pentyne was much less efficient (30% yield) and led to an inseparable 15 : 85 mixture of α/β vinylboronates (14, entry 12), showing a background steric regiocontrol favoring the borylation at the

least hindered alkyne terminus. Accordingly, the 3-heptyne, with a reduced steric bias, led to much poorer regiocontrol and similar low yield (43 : 57 mixture, 37%, entry 13).

Structural variation at the alkyl acetylenic substituent (R¹) was also found to be viable (Scheme 1). For example, alkyl

Scheme 1. Other Propargylic-Substituted Internal Alkynes^a



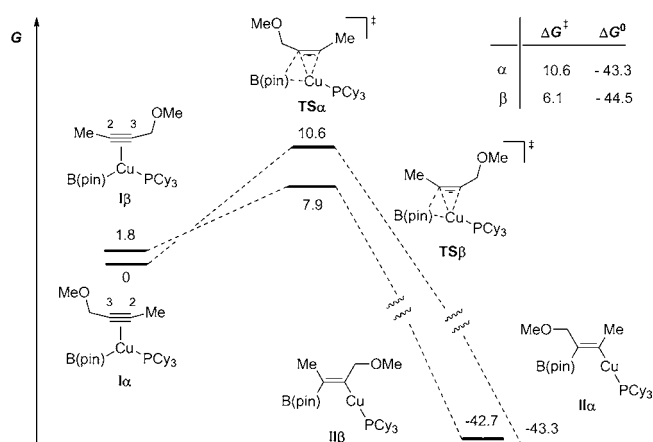
^aYield of β -regioisomer after chromatographic separation. ^bYield in the mixture of regioisomers.

groups bulkier than methyl such as propyl, isobutyl, or phenethyl are well accommodated, preserving high regiocontrol (products 16–18, from 93 to >98% β -selectivity).¹⁵ A cyanopropyl group could be installed with somewhat lower regioselectivity (19, 87% β). Similar β -selectivity (86%) was also observed in the borylation of 2-hexynol and its OTBS-protected derivative (products 20 and 21), as well as the 5-phenyl-2-pentynol (product 22, 84% β). Interestingly, the β -selectivity was restored to $\alpha/\beta = <2 : >98$ in the case of the bulkier secondary alcohol derivatives (R² = Me or Ph, products 23–25), suggesting the modulating role of steric effects in the main regiochemical control exerted by the propargylic moiety.¹⁶

Despite the borylation reaction was routinely run at 0.26 mmol scale, we confirmed that the process is amenable to 10-fold scale-up without loss of chemical efficiency, such as in the case of products 4 (2.05 mmol, 76%), 7 (2.50 mmol, 85%), and 17 (1.64 mmol, 92%). Additionally, we found that the Cu^I catalyst loading can be reduced to 3 mol % with comparable efficiency (product 17, 1.64 mmol, 87% yield, $\alpha/\beta = <2 : >98$).

DFT calculations provided insight into the reasons for the observed regioselectivity. According to the accepted mechanism, formation of the B–C bond takes place in the coordination sphere of a boryl-Cu complex formed by initial transmetalation. We focused on the boryl-cupration of 4-methoxy-2-butyne coordinated to (pin)B–Cu–PCy₃, considering the complete phosphine ligand structure. Optimizations were performed with CPCM model in toluene (see Supporting Information for details).¹⁷ Coordination of the alkyne may afford two different stereoisomeric trigonal complexes, **I α** and **I β** , leading to the alkenylboronate regioisomers **II α** and **II β** (Scheme 2). Both Cu–C(2) and Cu–C(3) distances are very similar in **I β** , in spite of the unsymmetrical nature of the alkyne (2.016 and 2.073 Å, respectively). In contrast, Cu–C(2) is significantly longer than Cu–C(3) in **I α** (2.187 and 2.032 Å, respectively), which is 1.8 kcal mol^{−1} more stable than **I β** .¹⁸

Location of both transition states (TS, Figure 1) for the insertion of the alkyne into the Cu–B bond allowed us to

Scheme 2. Energy Profile for the Two Possible Boryl Cuprations of the Coordinated Alkyne^a

^aOptimized structures in toluene (M06/6-31G(d) (C,H,B,O,P), LANL2DZ (Cu), CPCM model). Relative G values at 298 K (kcal mol⁻¹).

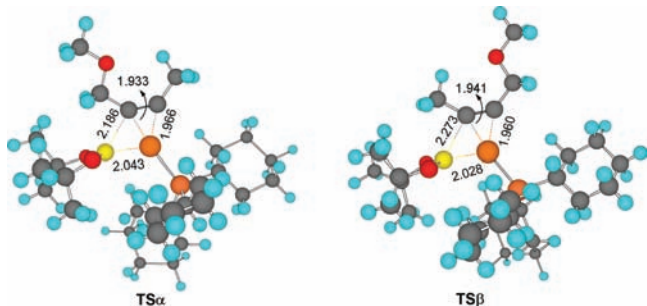
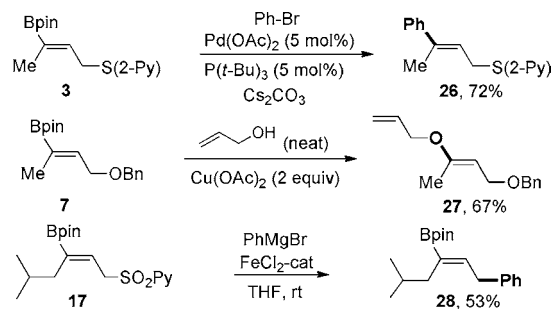


Figure 1. Molecular structure of the transition states **TS α** and **TS β** . Representative distances (Å) are indicated.

obtain the corresponding activation energies, which are low (6.1 and 10.6 kcal mol⁻¹ for β and α pathways, respectively) and in agreement with the experimentally observed regioselectivity. Thus, **TS β** lies 2.7 kcal mol⁻¹ below **TS α** . The carbon atom that binds to B strongly interacts with Cu in the TS, the Cu–C distances being shorter than those found in the starting complexes.¹⁹ The lower energy pathway involves formation of an earlier TS (**TS β**), with a longer C–B distance (2.273 vs 2.186 Å for **TS α**). NBO analysis reveals a strong donor–acceptor interaction between boron and the $\pi^*(C-C)$ alkyne orbital in both TSs. Interestingly, analysis of the HOMO of **TS β** shows that the O atom orbitals also participate in this MO, along with those of the alkyne and boryl moieties. Such participation is much smaller in **TS α** , suggesting that this effect could lay at the basis of the β -regiocontrol (see schemes in Supporting Information). Therefore, the main reason for the observed regioselectivity seems to be the result of orbital control rather than merely steric effects.^{20,21}

This formal hydroboration protocol offers an efficient approach to the stereocontrolled synthesis of multisubstituted alkenes²² via cross-coupling reactions of the resulting products (Scheme 3). For example, the Suzuki reaction of sulfide **3** with bromobenzene afforded the corresponding trisubstituted olefin **26** in 72% yield. The Cu-promoted coupling with alcohols, recently reported by the group of Merlic²³ on terminal vinylboronates, was found to be applicable to the branched vinylboronate **7**, leading to the allyl vinyl ether **27** (67% yield).

Scheme 3. Synthetic Applications of the Vinylboronates



Further elaboration of the products without affecting the boronate group via nucleophilic allylic displacement of the directing functionality allowed widening the scope of the vinylboronate synthesis. This concept was realized in the Fe^{II}-catalyzed allylic substitution of the 2-pyridylsulfone **17** with PhMgBr, affording the new vinylboronate **28** (53% yield, unoptimized, Scheme 3). Remarkably, this reaction shows the compatibility of the pinacolboronate group with Grignard reagents under metal catalysis²⁴ and, to the best of our knowledge, represents the first example of a Fe-catalyzed nucleophilic displacement of allylic sulfones with Grignard reagents.²⁵

In conclusion, we have disclosed a highly regiocontrolled B₂(pin)₂/Cu^I-catalyzed borylation of dialkylalkynes. The effectiveness of this method relies on the subtle electronic bias provided by a functional group at the propargylic position of the alkyne substrate, along with the use of the PCy₃ as ligand. Functionalized vinylboronates bearing a variety of sulfur, oxygen, and nitrogen groups at the allylic position are obtained in good yields and good to excellent β -selectivity. DFT calculations on the whole catalyst system are in agreement with the observed regioselectivity. The resulting products have been applied to the stereocontrolled synthesis of trisubstituted olefins via cross-coupling reactions. Further elaboration of the vinylboronates via Fe^{II}-catalyzed sulfonyl allylic substitution with Grignard reagents provides formal hydroboration products of unbiased dialkyl-substituted alkynes.

■ ASSOCIATED CONTENT

S Supporting Information

Experimental and computational details as well as spectroscopic and analytical data for new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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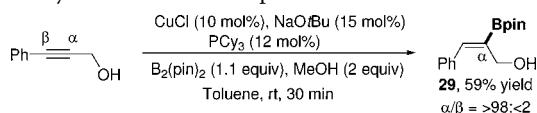
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